

Acta Crystallographica Section E

Structure Reports

Online

ISSN 1600-5368

Methyl 4-hydroxy-2-methoxycarbonyl-methyl-1,1-dioxo-1,2-dihydro-1 λ^6 ,2-benzothiazine-3-carboxylate¹Muhammad Nadeem Arshad,^{a,*}§ Islam Ullah Khan,^b Muhammad Zia-ur-Rehman,^c Sheikh Asrar Ahmad^d and H. M. Rafique^a

^aX-ray Diffraction and Crystallography Laboratory, Department of Physics, School of Physical Sciences, University of the Punjab, Quaid-e-Azam Campus, Lahore 54590, Pakistan, ^bMaterials Chemistry Laboratory, Department of Chemistry, GC University, Lahore 54000, Pakistan, ^cApplied Chemistry Research Centre, PCSIR Laboratories Complex, Lahore 54600, Pakistan, and ^dDepartment of Chemistry, Division of Science and Technology, University of Education, Lahore, Pakistan
Correspondence e-mail: mnachemist@hotmail.com

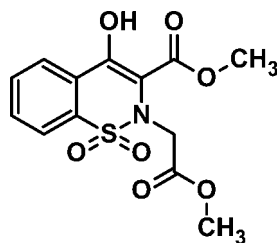
Received 24 August 2011; accepted 25 August 2011

Key indicators: single-crystal X-ray study; $T = 100$ K; mean $\sigma(\text{C}-\text{C}) = 0.002$ Å; R factor = 0.040; wR factor = 0.107; data-to-parameter ratio = 17.0.

There are two independent molecules in the asymmetric unit of the title compound, $\text{C}_{13}\text{H}_{13}\text{NO}_7\text{S}$, which have almost identical geometries. The thiazine ring adopts a sofa conformation in both molecules and the molecular conformations are stabilized by intramolecular $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds. Intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds stabilize the crystal packing.

Related literature

For related structures, see; Arshad *et al.* (2009, 2010). For the synthesis, see; Arshad *et al.* (2011).



Experimental

Crystal data

$\text{C}_{13}\text{H}_{13}\text{NO}_7\text{S}$
 $M_r = 327.30$

Triclinic, $P\bar{1}$
 $a = 8.9128$ (14) Å

$b = 12.414$ (2) Å
 $c = 13.443$ (2) Å
 $\alpha = 79.784$ (2)°
 $\beta = 72.981$ (3)°
 $\gamma = 88.503$ (3)°
 $V = 1399.2$ (4) Å³

$Z = 4$
Mo $K\alpha$ radiation
 $\mu = 0.27$ mm⁻¹
 $T = 100$ K
 $0.48 \times 0.36 \times 0.33$ mm

Data collection

Siemens SMART diffractometer equipped with a Bruker APEXII detector
Absorption correction: multi-scan (SADABS; Bruker, 2001)
 $T_{\min} = 0.882$, $T_{\max} = 0.917$

17331 measured reflections
6923 independent reflections
6322 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.018$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$
 $wR(F^2) = 0.107$
 $S = 1.06$
6923 reflections
407 parameters

H atoms treated by a mixture of independent and constrained refinement
 $\Delta\rho_{\text{max}} = 0.48$ e Å⁻³
 $\Delta\rho_{\text{min}} = -0.42$ e Å⁻³

Table 1

Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$\text{O1}-\text{H1O}\cdots\text{O4}$	0.79 (3)	1.87 (3)	2.5754 (18)	149 (3)
$\text{O8}-\text{H8O}\cdots\text{O11}$	0.98 (3)	1.68 (3)	2.5623 (17)	148 (3)
$\text{C13}-\text{H13A}\cdots\text{O10}^{\text{i}}$	0.98	2.44	3.202 (2)	134
$\text{C2}-\text{H2}\cdots\text{O11}^{\text{ii}}$	0.95	2.59	3.339 (2)	136

Symmetry codes: (i) $-x, -y + 2, -z + 1$; (ii) $-x, -y + 1, -z + 1$.

Data collection: APEX2 (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 2008); program(s) used to refine structure: SHELXL97 (Sheldrick, 2008); molecular graphics: PLATON (Spek, 2009) and X-SEED (Barbour, 2001); software used to prepare material for publication: WinGX (Farrugia, 1999).

MNA acknowledges the Higher Education Commission of Pakistan for granting a scholarship under its indigenous and IRSIP schemes.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: BT5629).

References

- Arshad, M. N., Khan, I. U., Zia-ur-Rehman, M. & Shafiq, M. (2011). *Asian J. Chem.* **23**, 2801–2805.
Arshad, M. N., Zia-ur-Rehman, M. & Khan, I. U. (2009). *Acta Cryst.* **E65**, o3077.
Arshad, M. N., Zia-ur-Rehman, M. & Khan, I. U. (2010). *Acta Cryst.* **E66**, o1070.
Barbour, L. J. (2001). *J. Supramol. Chem.* **1**, 189–191.
Bruker (2001). SADABS, APEX2 and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.
Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
Sheldrick, G. M. (2008). *Acta Cryst.* **A64**, 112–122.
Spek, A. L. (2009). *Acta Cryst.* **D65**, 148–155.

¹ I (MNA) dedicate this article to my first child (a boy), Mr Muhammad Ali Hassan, as he was born on 22nd August 2011, the day I compiled this article. § Materials Chemistry Laboratory, Department of Chemistry, GC University, Lahore 54000, Pakistan.

supplementary materials

Acta Cryst. (2011). E67, o2527 [doi:10.1107/S1600536811034921]

Methyl 4-hydroxy-2-methoxycarbonylmethyl-1,1-dioxo-1,2-dihydro-1 λ 6,2-benzothiazine-3-carboxylate

M. N. Arshad, I. U. Khan, M. Zia-ur-Rehman, S. A. Ahmad and H. M. Rafique

Comment

The crystal structure of title compound is being reported in continuation of our structural studies of various benzothiazines (Arshad *et al.*, 2009, 2010).

The title compound crystallizes with two independent molecules in the asymmetric unit (Fig. 1). In molecule A, the two fused rings are oriented at dihedral angle of 14.18 (7) $^{\circ}$ while in B this dihedral angle amounts to 18.40 (6) $^{\circ}$. The root mean square deviation values for thiazine ring in both molecules are 0.1991 \AA and 0.2119 \AA respectively. The alkylated groups attached to the nitrogen atom in molecule A and B are inclined at almost perpendicular position with respect to the thiazine rings and given values are 85.81 (5) $^{\circ}$ and 88.12 (5) $^{\circ}$. Both of the molecules are involved in intra and intermolecular hydrogen bondings. The intramolecular hydrogen bonding of O—H \cdots O type forms six membered ring motif in each molecule. Other inter molecular C—H \cdots O and O—H \cdots O type hydrogen bonding interactions produce three dimensional network (Fig. 2 and Tab. 1).

Experimental

The synthesis of title compound is already reported (Arshad *et al.*, 2011). The compound was recrystallized under slow evaporation technique in ethylacetate.

Refinement

The C—H atoms were positioned with idealized geometry with C—H = 0.95 \AA for aromatic, C—H = 0.99 \AA for methylene & C—H = 0.98 \AA for methyl groups and were refined using a riding model with $U_{\text{iso}}(\text{H}) = 1.2 U_{\text{eq}}(\text{C})$ for aromatic & methylene and $U_{\text{iso}}(\text{H}) = 1.5 U_{\text{eq}}(\text{C})$ for methyl groups.

The O—H atoms were located in a difference map and refined isotropically.

Figures

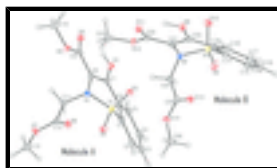


Fig. 1. The *ORTEP* diagram of the title compound with displacement ellipsoids drawn at the 30% probability level.

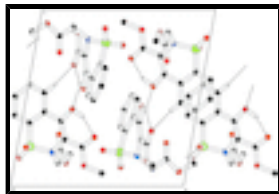


Fig. 2. Unit cell packing of the title compound showing hydrogen bonding interaction as dashed lines.

Methyl 4-hydroxy-2-methoxycarbonylmethyl-1,1-dioxo-1,2-dihydro-1 λ ⁶,2-benzothiazine-3-carboxylate

Crystal data

$C_{13}H_{13}NO_7S$	$Z = 4$
$M_r = 327.30$	$F(000) = 680$
Triclinic, $P\bar{1}$	$D_x = 1.554 \text{ Mg m}^{-3}$
Hall symbol: $-P\ 1$	Mo $K\alpha$ radiation, $\lambda = 0.71073 \text{ \AA}$
$a = 8.9128 (14) \text{ \AA}$	Cell parameters from 9730 reflections
$b = 12.414 (2) \text{ \AA}$	$\theta = 2.4\text{--}28.9^\circ$
$c = 13.443 (2) \text{ \AA}$	$\mu = 0.27 \text{ mm}^{-1}$
$\alpha = 79.784 (2)^\circ$	$T = 100 \text{ K}$
$\beta = 72.981 (3)^\circ$	Blocks, colorless
$\gamma = 88.503 (3)^\circ$	$0.48 \times 0.36 \times 0.33 \text{ mm}$
$V = 1399.2 (4) \text{ \AA}^3$	

Data collection

Siemens SMART diffractometer equipped with a Bruker APEXII detector	6923 independent reflections
Radiation source: fine-focus sealed tube graphite	6322 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.018$
Absorption correction: multi-scan (SADABS; Bruker, 2001)	$\theta_{\text{max}} = 28.9^\circ$, $\theta_{\text{min}} = 2.1^\circ$
$T_{\text{min}} = 0.882$, $T_{\text{max}} = 0.917$	$h = -12 \rightarrow 12$
17331 measured reflections	$k = -16 \rightarrow 16$
	$l = -18 \rightarrow 17$

Refinement

Refinement on F^2	Primary atom site location: structure-invariant direct methods
Least-squares matrix: full	Secondary atom site location: difference Fourier map
$R[F^2 > 2\sigma(F^2)] = 0.040$	Hydrogen site location: inferred from neighbouring sites
$wR(F^2) = 0.107$	H atoms treated by a mixture of independent and constrained refinement
$S = 1.06$	$w = 1/[\sigma^2(F_o^2) + (0.0516P)^2 + 0.9585P]$
6923 reflections	where $P = (F_o^2 + 2F_c^2)/3$
	$(\Delta/\sigma)_{\text{max}} = 0.001$

407 parameters

$$\Delta\rho_{\max} = 0.48 \text{ e } \text{\AA}^{-3}$$

0 restraints

$$\Delta\rho_{\min} = -0.42 \text{ e } \text{\AA}^{-3}$$

Special details

Geometry. All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted R-factor wR and goodness of fit S are based on F^2 , conventional R-factors R are based on F, with F set to zero for negative F^2 . The threshold expression of $F^2 > 2\sigma(F^2)$ is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on F^2 are statistically about twice as large as those based on F, and R- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
S1	0.16821 (4)	0.59653 (3)	0.19082 (3)	0.01849 (9)
S2	0.52552 (5)	0.92829 (3)	0.77528 (3)	0.02196 (10)
O8	0.48743 (15)	0.72159 (10)	0.56519 (9)	0.0263 (3)
O1	0.30206 (15)	0.68027 (10)	0.44663 (9)	0.0240 (2)
O12	0.38864 (14)	0.59665 (9)	0.88368 (9)	0.0225 (2)
O4	0.56249 (14)	0.75774 (10)	0.31183 (9)	0.0267 (3)
O7	0.12303 (14)	0.95735 (10)	0.08091 (9)	0.0269 (3)
O11	0.39383 (14)	0.56505 (10)	0.72314 (9)	0.0259 (2)
O14	0.89610 (14)	0.70395 (10)	0.86890 (10)	0.0276 (3)
O3	0.25344 (14)	0.49769 (9)	0.20325 (9)	0.0245 (2)
N1	0.29039 (15)	0.70033 (10)	0.17185 (10)	0.0181 (2)
O13	0.86686 (14)	0.77263 (11)	0.70832 (10)	0.0286 (3)
O2	0.09291 (14)	0.61619 (10)	0.10870 (9)	0.0236 (2)
O5	0.59329 (13)	0.76066 (10)	0.13910 (9)	0.0229 (2)
N2	0.53834 (15)	0.79462 (11)	0.80675 (10)	0.0179 (2)
O9	0.59840 (17)	0.97841 (11)	0.83798 (10)	0.0343 (3)
O6	0.06605 (15)	0.85459 (10)	0.24309 (9)	0.0275 (3)
O10	0.36468 (15)	0.94991 (11)	0.78104 (11)	0.0342 (3)
C19	0.61858 (19)	0.87842 (13)	0.58043 (12)	0.0211 (3)
C11	0.27512 (18)	0.80160 (13)	0.10098 (12)	0.0202 (3)
H11A	0.2567	0.7827	0.0369	0.024*
H11B	0.3751	0.8446	0.0785	0.024*
C3	-0.2273 (2)	0.57856 (14)	0.43419 (13)	0.0249 (3)
H3	-0.3349	0.5583	0.4499	0.030*
C14	0.63731 (19)	0.95418 (13)	0.64225 (12)	0.0203 (3)
C9	0.50907 (19)	0.74369 (12)	0.24092 (12)	0.0201 (3)
C8	0.34682 (18)	0.70707 (12)	0.26034 (12)	0.0189 (3)
C20	0.53123 (18)	0.77482 (13)	0.63114 (12)	0.0200 (3)
C6	0.08927 (18)	0.64003 (12)	0.38905 (12)	0.0188 (3)
C1	0.03184 (18)	0.60509 (12)	0.31319 (12)	0.0187 (3)

supplementary materials

C2	-0.12533 (19)	0.57587 (13)	0.33394 (13)	0.0215 (3)
H2	-0.1625	0.5545	0.2807	0.026*
C15	0.7366 (2)	1.04599 (13)	0.60003 (14)	0.0269 (3)
H15	0.7499	1.0952	0.6436	0.032*
C5	-0.01566 (19)	0.64217 (12)	0.48944 (12)	0.0216 (3)
H5	0.0204	0.6651	0.5424	0.026*
C22	0.42446 (18)	0.62469 (13)	0.77922 (12)	0.0204 (3)
C24	0.64439 (19)	0.75062 (14)	0.86710 (12)	0.0229 (3)
H24A	0.6064	0.6760	0.9049	0.027*
H24B	0.6399	0.7964	0.9211	0.027*
C21	0.50090 (17)	0.73286 (12)	0.73654 (12)	0.0184 (3)
C18	0.6976 (2)	0.89909 (15)	0.47193 (13)	0.0302 (4)
H18	0.6846	0.8501	0.4280	0.036*
C12	0.14236 (18)	0.87217 (13)	0.15144 (12)	0.0207 (3)
C25	0.81345 (18)	0.74493 (13)	0.80252 (13)	0.0209 (3)
C4	-0.1721 (2)	0.61080 (13)	0.51154 (13)	0.0246 (3)
H4	-0.2422	0.6113	0.5800	0.029*
C23	0.3242 (2)	0.48585 (14)	0.92729 (14)	0.0303 (4)
H23A	0.2267	0.4773	0.9093	0.045*
H23B	0.3021	0.4727	1.0043	0.045*
H23C	0.4002	0.4330	0.8976	0.045*
C10	0.7541 (2)	0.79979 (16)	0.11653 (14)	0.0290 (4)
H10A	0.7569	0.8799	0.1072	0.043*
H10B	0.8190	0.7768	0.0517	0.043*
H10C	0.7949	0.7690	0.1755	0.043*
C7	0.25286 (19)	0.67701 (12)	0.36224 (12)	0.0195 (3)
C17	0.7950 (3)	0.99176 (17)	0.42903 (14)	0.0366 (4)
H17	0.8479	1.0059	0.3554	0.044*
C13	0.0082 (2)	1.03446 (15)	0.12689 (15)	0.0309 (4)
H13A	-0.0919	0.9955	0.1652	0.046*
H13B	-0.0070	1.0915	0.0705	0.046*
H13C	0.0464	1.0681	0.1759	0.046*
C16	0.8164 (3)	1.06409 (15)	0.49221 (15)	0.0348 (4)
H16	0.8855	1.1260	0.4618	0.042*
C26	1.0610 (2)	0.6897 (2)	0.8203 (2)	0.0437 (5)
H26A	1.0733	0.6389	0.7705	0.066*
H26B	1.1107	0.6600	0.8750	0.066*
H26C	1.1108	0.7606	0.7823	0.066*
H10	0.389 (4)	0.703 (3)	0.425 (2)	0.066*
H80	0.437 (4)	0.653 (3)	0.608 (2)	0.066*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
S1	0.01975 (18)	0.01908 (18)	0.01976 (18)	0.00083 (13)	-0.00782 (14)	-0.00819 (13)
S2	0.0236 (2)	0.02065 (19)	0.02086 (19)	0.00131 (14)	-0.00283 (14)	-0.00805 (14)
O8	0.0324 (6)	0.0306 (6)	0.0196 (5)	-0.0077 (5)	-0.0116 (5)	-0.0059 (5)
O1	0.0298 (6)	0.0259 (6)	0.0195 (5)	-0.0054 (5)	-0.0127 (5)	-0.0018 (4)

O12	0.0272 (6)	0.0190 (5)	0.0187 (5)	-0.0014 (4)	-0.0031 (4)	-0.0024 (4)
O4	0.0269 (6)	0.0318 (6)	0.0258 (6)	-0.0035 (5)	-0.0129 (5)	-0.0069 (5)
O7	0.0287 (6)	0.0288 (6)	0.0216 (5)	0.0065 (5)	-0.0076 (5)	-0.0012 (5)
O11	0.0279 (6)	0.0255 (6)	0.0253 (6)	-0.0050 (5)	-0.0079 (5)	-0.0062 (5)
O14	0.0231 (6)	0.0296 (6)	0.0337 (6)	0.0044 (5)	-0.0146 (5)	-0.0047 (5)
O3	0.0268 (6)	0.0202 (5)	0.0302 (6)	0.0040 (4)	-0.0102 (5)	-0.0111 (5)
N1	0.0196 (6)	0.0197 (6)	0.0178 (6)	-0.0005 (5)	-0.0083 (5)	-0.0055 (5)
O13	0.0232 (6)	0.0343 (7)	0.0254 (6)	0.0027 (5)	-0.0037 (5)	-0.0036 (5)
O2	0.0236 (6)	0.0301 (6)	0.0216 (5)	-0.0004 (4)	-0.0102 (4)	-0.0097 (5)
O5	0.0193 (5)	0.0272 (6)	0.0239 (5)	-0.0013 (4)	-0.0064 (4)	-0.0088 (4)
N2	0.0194 (6)	0.0204 (6)	0.0160 (6)	0.0000 (5)	-0.0075 (5)	-0.0048 (5)
O9	0.0453 (8)	0.0330 (7)	0.0241 (6)	-0.0119 (6)	-0.0030 (5)	-0.0142 (5)
O6	0.0313 (6)	0.0231 (6)	0.0235 (6)	0.0025 (5)	-0.0023 (5)	-0.0025 (4)
O10	0.0265 (6)	0.0325 (7)	0.0385 (7)	0.0104 (5)	-0.0024 (5)	-0.0063 (6)
C19	0.0250 (8)	0.0216 (7)	0.0185 (7)	-0.0011 (6)	-0.0098 (6)	-0.0021 (6)
C11	0.0222 (7)	0.0225 (7)	0.0167 (7)	0.0009 (6)	-0.0068 (6)	-0.0039 (5)
C3	0.0215 (8)	0.0227 (8)	0.0288 (8)	0.0004 (6)	-0.0061 (6)	-0.0019 (6)
C14	0.0233 (7)	0.0194 (7)	0.0182 (7)	0.0026 (6)	-0.0065 (6)	-0.0031 (5)
C9	0.0230 (7)	0.0169 (7)	0.0227 (7)	0.0013 (5)	-0.0091 (6)	-0.0054 (5)
C8	0.0222 (7)	0.0186 (7)	0.0196 (7)	0.0008 (5)	-0.0099 (6)	-0.0064 (5)
C20	0.0200 (7)	0.0233 (7)	0.0194 (7)	-0.0008 (6)	-0.0081 (6)	-0.0063 (6)
C6	0.0236 (7)	0.0143 (6)	0.0195 (7)	0.0002 (5)	-0.0079 (6)	-0.0028 (5)
C1	0.0226 (7)	0.0155 (6)	0.0189 (7)	0.0012 (5)	-0.0068 (6)	-0.0044 (5)
C2	0.0228 (7)	0.0187 (7)	0.0251 (7)	0.0000 (6)	-0.0097 (6)	-0.0043 (6)
C15	0.0349 (9)	0.0193 (8)	0.0265 (8)	-0.0023 (6)	-0.0085 (7)	-0.0045 (6)
C5	0.0293 (8)	0.0178 (7)	0.0183 (7)	0.0008 (6)	-0.0080 (6)	-0.0028 (5)
C22	0.0167 (7)	0.0230 (7)	0.0216 (7)	0.0013 (5)	-0.0053 (6)	-0.0050 (6)
C24	0.0210 (7)	0.0327 (8)	0.0167 (7)	0.0006 (6)	-0.0090 (6)	-0.0031 (6)
C21	0.0190 (7)	0.0205 (7)	0.0176 (7)	-0.0013 (5)	-0.0068 (5)	-0.0056 (5)
C18	0.0431 (10)	0.0297 (9)	0.0182 (7)	-0.0068 (7)	-0.0091 (7)	-0.0037 (6)
C12	0.0211 (7)	0.0199 (7)	0.0235 (7)	-0.0006 (6)	-0.0102 (6)	-0.0041 (6)
C25	0.0210 (7)	0.0185 (7)	0.0249 (7)	0.0009 (5)	-0.0090 (6)	-0.0045 (6)
C4	0.0269 (8)	0.0220 (8)	0.0219 (7)	0.0033 (6)	-0.0038 (6)	-0.0022 (6)
C23	0.0421 (10)	0.0170 (8)	0.0257 (8)	-0.0006 (7)	-0.0024 (7)	-0.0003 (6)
C10	0.0202 (8)	0.0378 (10)	0.0303 (9)	-0.0047 (7)	-0.0060 (6)	-0.0111 (7)
C7	0.0259 (8)	0.0154 (7)	0.0205 (7)	-0.0001 (5)	-0.0110 (6)	-0.0041 (5)
C17	0.0505 (12)	0.0348 (10)	0.0198 (8)	-0.0098 (8)	-0.0043 (8)	-0.0015 (7)
C13	0.0314 (9)	0.0281 (9)	0.0316 (9)	0.0054 (7)	-0.0085 (7)	-0.0035 (7)
C16	0.0465 (11)	0.0255 (9)	0.0272 (9)	-0.0110 (8)	-0.0048 (8)	0.0004 (7)
C26	0.0275 (10)	0.0493 (12)	0.0578 (13)	0.0141 (9)	-0.0168 (9)	-0.0134 (10)

Geometric parameters (Å, °)

S1—O2	1.4347 (12)	C14—C15	1.391 (2)
S1—O3	1.4365 (12)	C9—O4	1.2226 (19)
S1—N1	1.6449 (13)	C9—C8	1.462 (2)
S1—C1	1.7548 (16)	C8—C7	1.371 (2)
S2—O9	1.4308 (13)	C20—O8	1.3413 (18)
S2—O10	1.4337 (14)	C20—C21	1.369 (2)

supplementary materials

S2—N2	1.6477 (14)	C6—C1	1.401 (2)
S2—C14	1.7504 (16)	C6—C5	1.403 (2)
O8—C20	1.3413 (18)	C6—C7	1.461 (2)
O8—H8O	0.98 (3)	C1—C2	1.391 (2)
O1—C7	1.3377 (18)	C2—H2	0.9500
O1—H10	0.79 (3)	C15—C16	1.395 (2)
O12—C22	1.3285 (19)	C15—H15	0.9500
O12—C23	1.457 (2)	C5—C4	1.389 (2)
O4—C9	1.2226 (19)	C5—H5	0.9500
O7—C12	1.3330 (19)	C22—C21	1.464 (2)
O7—C13	1.461 (2)	C24—C25	1.510 (2)
O11—C22	1.2329 (19)	C24—H24A	0.9900
O14—C25	1.3423 (19)	C24—H24B	0.9900
O14—C26	1.443 (2)	C18—C17	1.391 (3)
N1—C8	1.4362 (18)	C18—H18	0.9500
N1—C11	1.4658 (19)	C4—H4	0.9500
O13—C25	1.204 (2)	C23—H23A	0.9800
O5—C9	1.3364 (19)	C23—H23B	0.9800
O5—C10	1.4535 (19)	C23—H23C	0.9800
N2—C21	1.4298 (18)	C10—H10A	0.9800
N2—C24	1.4543 (19)	C10—H10B	0.9800
O6—C12	1.204 (2)	C10—H10C	0.9800
C19—C18	1.402 (2)	C17—C16	1.392 (3)
C19—C14	1.403 (2)	C17—H17	0.9500
C19—C20	1.468 (2)	C13—H13A	0.9800
C11—C12	1.521 (2)	C13—H13B	0.9800
C11—H11A	0.9900	C13—H13C	0.9800
C11—H11B	0.9900	C16—H16	0.9500
C3—C4	1.393 (2)	C26—H26A	0.9800
C3—C2	1.393 (2)	C26—H26B	0.9800
C3—H3	0.9500	C26—H26C	0.9800
O2—S1—O3	118.97 (7)	C14—C15—H15	120.8
O2—S1—N1	107.65 (7)	C16—C15—H15	120.8
O3—S1—N1	107.63 (7)	C4—C5—C6	120.16 (15)
O2—S1—C1	110.07 (7)	C4—C5—H5	119.9
O3—S1—C1	108.17 (7)	C6—C5—H5	119.9
N1—S1—C1	103.18 (7)	O11—C22—O12	122.69 (14)
O9—S2—O10	119.43 (9)	O11—C22—C21	122.86 (14)
O9—S2—N2	107.81 (8)	O12—C22—C21	114.45 (13)
O10—S2—N2	107.01 (7)	N2—C24—C25	114.97 (13)
O9—S2—C14	110.62 (8)	N2—C24—H24A	108.5
O10—S2—C14	108.43 (8)	C25—C24—H24A	108.5
N2—S2—C14	102.09 (7)	N2—C24—H24B	108.5
C20—O8—H8O	106.3 (17)	C25—C24—H24B	108.5
C7—O1—H10	107 (2)	H24A—C24—H24B	107.5
C22—O12—C23	115.47 (13)	C20—C21—N2	120.96 (13)
C12—O7—C13	113.54 (13)	C20—C21—C22	119.85 (13)
C25—O14—C26	115.71 (15)	N2—C21—C22	119.16 (13)
C8—N1—C11	118.88 (12)	C17—C18—C19	119.55 (16)

C8—N1—S1	114.54 (10)	C17—C18—H18	120.2
C11—N1—S1	119.24 (10)	C19—C18—H18	120.2
C9—O5—C10	115.65 (12)	O6—C12—O7	124.32 (15)
C21—N2—C24	119.88 (13)	O6—C12—C11	124.73 (14)
C21—N2—S2	114.97 (10)	O7—C12—C11	110.94 (13)
C24—N2—S2	119.48 (11)	O13—C25—O14	125.30 (15)
C18—C19—C14	118.56 (15)	O13—C25—C24	126.78 (14)
C18—C19—C20	121.16 (14)	O14—C25—C24	107.92 (13)
C14—C19—C20	119.98 (14)	C5—C4—C3	120.59 (15)
N1—C11—C12	113.26 (13)	C5—C4—H4	119.7
N1—C11—H11A	108.9	C3—C4—H4	119.7
C12—C11—H11A	108.9	O12—C23—H23A	109.5
N1—C11—H11B	108.9	O12—C23—H23B	109.5
C12—C11—H11B	108.9	H23A—C23—H23B	109.5
H11A—C11—H11B	107.7	O12—C23—H23C	109.5
C4—C3—C2	120.26 (15)	H23A—C23—H23C	109.5
C4—C3—H3	119.9	H23B—C23—H23C	109.5
C2—C3—H3	119.9	O5—C10—H10A	109.5
C15—C14—C19	122.05 (15)	O5—C10—H10B	109.5
C15—C14—S2	121.66 (12)	H10A—C10—H10B	109.5
C19—C14—S2	116.28 (12)	O5—C10—H10C	109.5
O4—C9—O5	123.11 (14)	H10A—C10—H10C	109.5
O4—C9—O5	123.11 (14)	H10B—C10—H10C	109.5
O4—C9—C8	122.76 (14)	O1—C7—C8	123.00 (14)
O4—C9—C8	122.76 (14)	O1—C7—C6	113.63 (13)
O5—C9—C8	114.13 (13)	C8—C7—C6	123.36 (14)
C7—C8—N1	121.26 (14)	C18—C17—C16	121.10 (17)
C7—C8—C9	119.61 (13)	C18—C17—H17	119.4
N1—C8—C9	119.10 (13)	C16—C17—H17	119.4
O8—C20—C21	121.98 (14)	O7—C13—H13A	109.5
O8—C20—C21	121.98 (14)	O7—C13—H13B	109.5
O8—C20—C19	114.68 (13)	H13A—C13—H13B	109.5
O8—C20—C19	114.68 (13)	O7—C13—H13C	109.5
C21—C20—C19	123.29 (14)	H13A—C13—H13C	109.5
C1—C6—C5	118.23 (14)	H13B—C13—H13C	109.5
C1—C6—C7	120.43 (14)	C17—C16—C15	120.20 (17)
C5—C6—C7	121.30 (14)	C17—C16—H16	119.9
C2—C1—C6	121.94 (14)	C15—C16—H16	119.9
C2—C1—S1	120.99 (12)	O14—C26—H26A	109.5
C6—C1—S1	117.05 (12)	O14—C26—H26B	109.5
C1—C2—C3	118.78 (15)	H26A—C26—H26B	109.5
C1—C2—H2	120.6	O14—C26—H26C	109.5
C3—C2—H2	120.6	H26A—C26—H26C	109.5
C14—C15—C16	118.48 (16)	H26B—C26—H26C	109.5
O2—S1—N1—C8	165.62 (10)	N1—S1—C1—C2	145.82 (13)
O3—S1—N1—C8	-65.00 (12)	O2—S1—C1—C6	-150.26 (12)
C1—S1—N1—C8	49.24 (12)	O3—S1—C1—C6	78.24 (13)
O2—S1—N1—C11	15.84 (13)	N1—S1—C1—C6	-35.61 (13)
O3—S1—N1—C11	145.22 (11)	C6—C1—C2—C3	-1.9 (2)

supplementary materials

C1—S1—N1—C11	-100.54 (12)	S1—C1—C2—C3	176.63 (12)
O9—S2—N2—C21	-167.55 (11)	C4—C3—C2—C1	0.5 (2)
O10—S2—N2—C21	62.81 (12)	C19—C14—C15—C16	1.7 (3)
C14—S2—N2—C21	-50.99 (12)	S2—C14—C15—C16	-176.97 (15)
O9—S2—N2—C24	-14.11 (14)	C1—C6—C5—C4	-0.4 (2)
O10—S2—N2—C24	-143.74 (12)	C7—C6—C5—C4	177.35 (14)
C14—S2—N2—C24	102.45 (12)	C23—O12—C22—O11	5.7 (2)
C8—N1—C11—C12	-70.27 (17)	C23—O12—C22—C21	-175.20 (14)
S1—N1—C11—C12	78.20 (15)	C21—N2—C24—C25	69.53 (18)
C18—C19—C14—C15	-2.8 (3)	S2—N2—C24—C25	-82.61 (16)
C20—C19—C14—C15	170.92 (15)	O8—C20—C21—N2	-175.35 (14)
C18—C19—C14—S2	175.91 (13)	O8—C20—C21—N2	-175.35 (14)
C20—C19—C14—S2	-10.3 (2)	C19—C20—C21—N2	7.1 (2)
O9—S2—C14—C15	-25.92 (17)	O8—C20—C21—C22	2.7 (2)
O10—S2—C14—C15	106.82 (15)	O8—C20—C21—C22	2.7 (2)
N2—S2—C14—C15	-140.43 (14)	C19—C20—C21—C22	-174.91 (14)
O9—S2—C14—C19	155.31 (13)	C24—N2—C21—C20	-121.58 (16)
O10—S2—C14—C19	-71.94 (14)	S2—N2—C21—C20	31.75 (19)
N2—S2—C14—C19	40.81 (14)	C24—N2—C21—C22	60.39 (19)
O4—O4—C9—O5	0.00 (13)	S2—N2—C21—C22	-146.28 (12)
O4—O4—C9—C8	0.00 (6)	O11—C22—C21—C20	4.7 (2)
C10—O5—C9—O4	-1.3 (2)	O12—C22—C21—C20	-174.38 (14)
C10—O5—C9—O4	-1.3 (2)	O11—C22—C21—N2	-177.28 (14)
C10—O5—C9—C8	178.51 (14)	O12—C22—C21—N2	3.7 (2)
C11—N1—C8—C7	114.97 (16)	C14—C19—C18—C17	1.7 (3)
S1—N1—C8—C7	-34.92 (18)	C20—C19—C18—C17	-171.98 (18)
C11—N1—C8—C9	-66.85 (18)	C13—O7—C12—O6	5.0 (2)
S1—N1—C8—C9	143.26 (12)	C13—O7—C12—C11	-174.02 (13)
O4—C9—C8—C7	-6.3 (2)	N1—C11—C12—O6	7.2 (2)
O4—C9—C8—C7	-6.3 (2)	N1—C11—C12—O7	-173.69 (12)
O5—C9—C8—C7	173.89 (13)	C26—O14—C25—O13	-2.1 (2)
O4—C9—C8—N1	175.49 (14)	C26—O14—C25—C24	178.36 (15)
O4—C9—C8—N1	175.49 (14)	N2—C24—C25—O13	-0.1 (2)
O5—C9—C8—N1	-4.3 (2)	N2—C24—C25—O14	179.43 (13)
O8—O8—C20—C21	0.00 (15)	C6—C5—C4—C3	-1.0 (2)
O8—O8—C20—C19	0.00 (19)	C2—C3—C4—C5	1.0 (2)
C18—C19—C20—O8	-22.0 (2)	N1—C8—C7—O1	178.77 (13)
C14—C19—C20—O8	164.35 (14)	C9—C8—C7—O1	0.6 (2)
C18—C19—C20—O8	-22.0 (2)	N1—C8—C7—C6	-2.2 (2)
C14—C19—C20—O8	164.35 (14)	C9—C8—C7—C6	179.64 (14)
C18—C19—C20—C21	155.70 (17)	C1—C6—C7—O1	-163.40 (14)
C14—C19—C20—C21	-17.9 (2)	C5—C6—C7—O1	18.9 (2)
C5—C6—C1—C2	1.8 (2)	C1—C6—C7—C8	17.5 (2)
C7—C6—C1—C2	-175.91 (14)	C5—C6—C7—C8	-160.19 (15)
C5—C6—C1—S1	-176.74 (11)	C19—C18—C17—C16	0.5 (3)
C7—C6—C1—S1	5.53 (19)	C18—C17—C16—C15	-1.6 (3)
O2—S1—C1—C2	31.17 (15)	C14—C15—C16—C17	0.5 (3)
O3—S1—C1—C2	-100.33 (14)		

Hydrogen-bond geometry (Å, °)

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> ··· <i>A</i>	<i>D</i> —H··· <i>A</i>
O1—H10···O4	0.79 (3)	1.87 (3)	2.5754 (18)	149 (3)
O8—H8O···O11	0.98 (3)	1.68 (3)	2.5623 (17)	148 (3)
C13—H13A···O10 ⁱ	0.98	2.44	3.202 (2)	134.
C2—H2···O11 ⁱⁱ	0.95	2.59	3.339 (2)	136.

Symmetry codes: (i) $-x, -y+2, -z+1$; (ii) $-x, -y+1, -z+1$.

Fig. 1

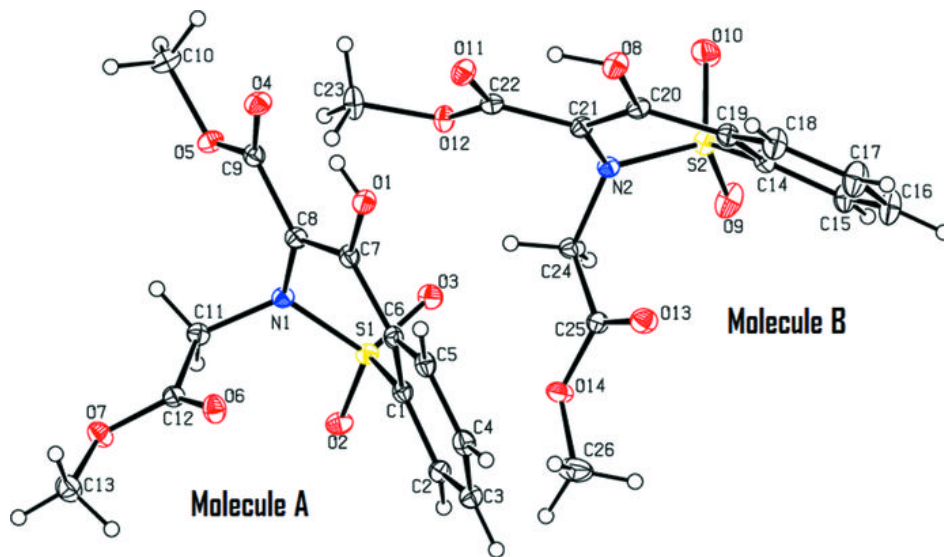


Fig. 2

